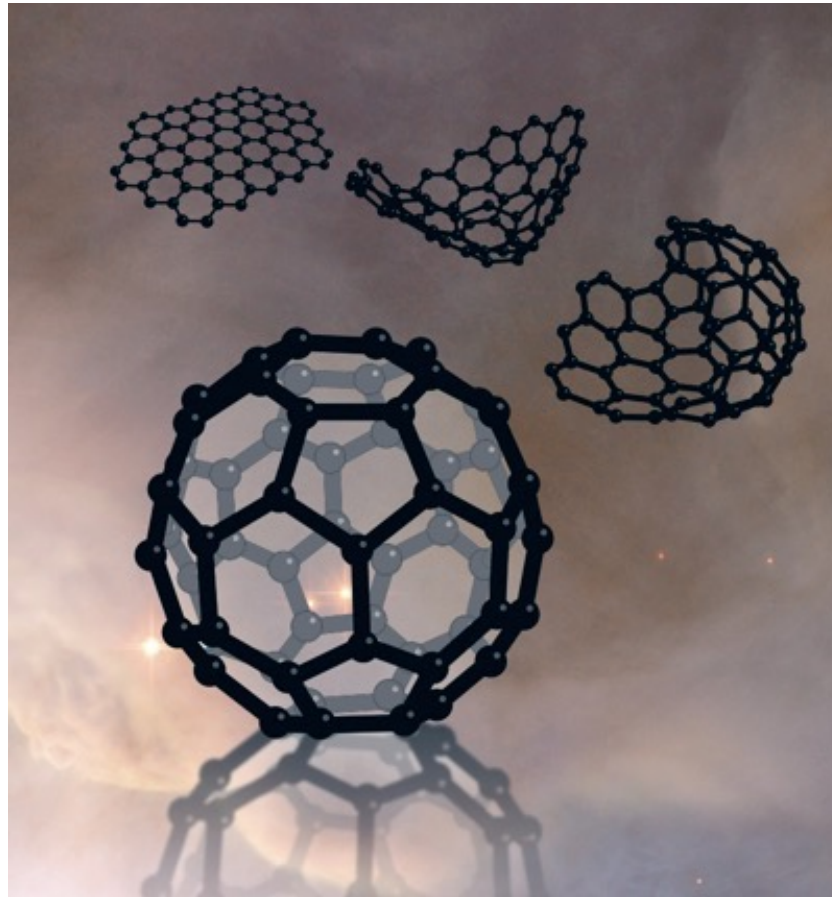


Cosmic fullerenes



O. Berné (IRAP / CNRS and Univ. Toulouse)

Contributors to the work presented today:

J. Montillaud (Obs. Besançon)

C. Joblin (IRAP / CNRS and Univ Toulouse)

G. Mulas (INAF Cagliari)

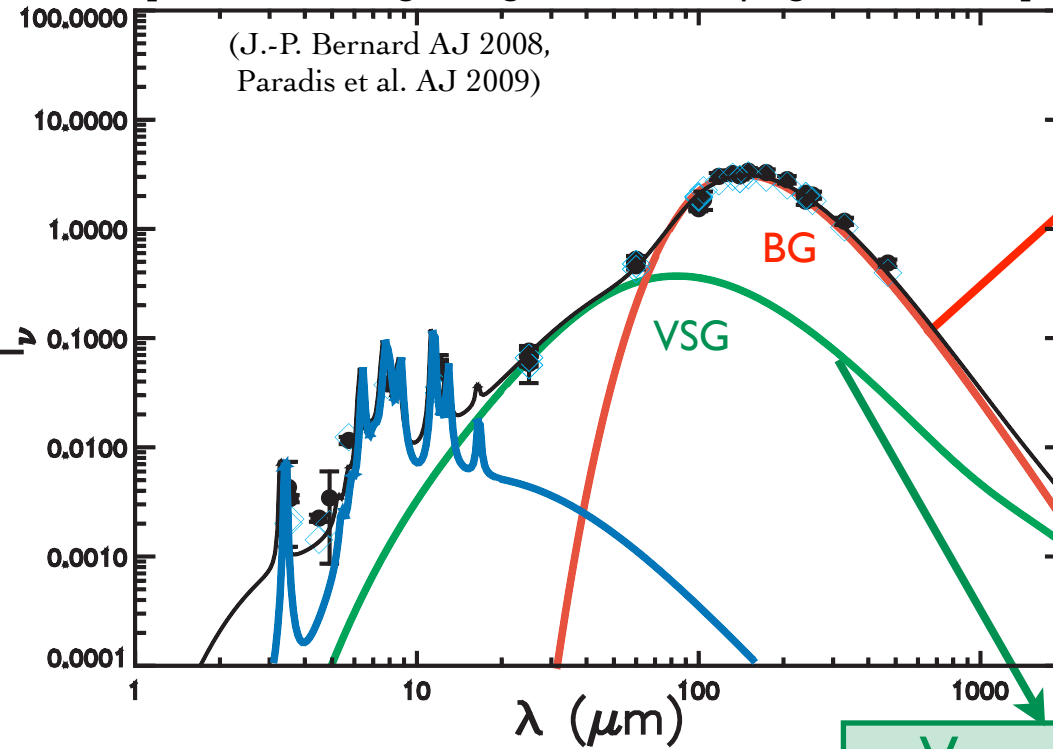
A. G. G. M. Tielens (Leiden University)

M. Rosenberg (Leiden University) **C. Boersma** (NASA Ames)

The infrared emission of dust in galaxies

[Désert, Boulanger, Puget 1990, Compiègne et al. 2011]

(J.-P. Bernard AJ 2008,
Paradis et al. AJ 2009)



Big Grains (BG)

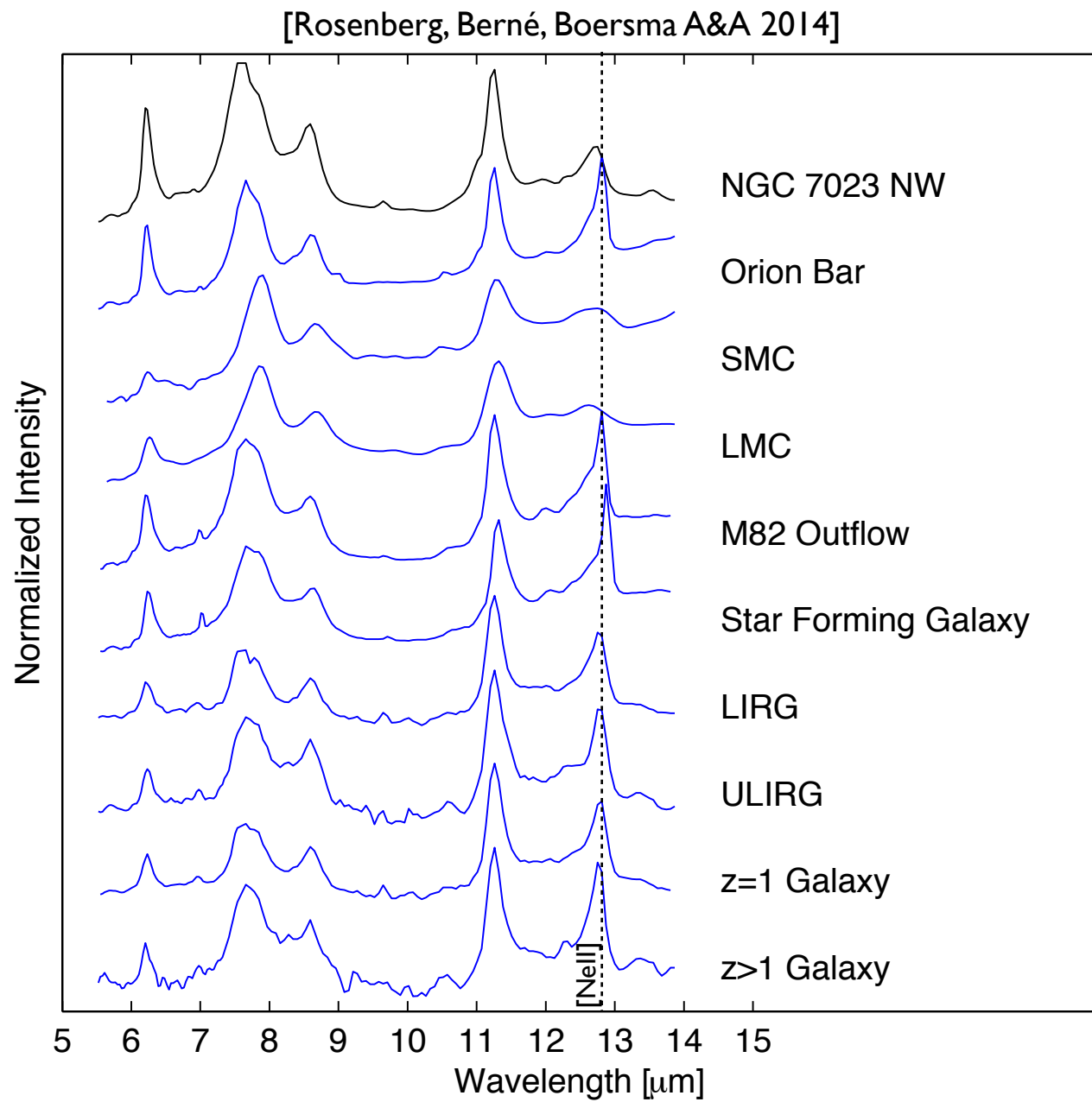
Silicate
size $> 0.1 \mu\text{m}$

Very Small Grains (VSGs)

carbonaceous,
10-100 nm...

(Désert et al. A&A, 1990; Draine & Li, ApJ, 2001)

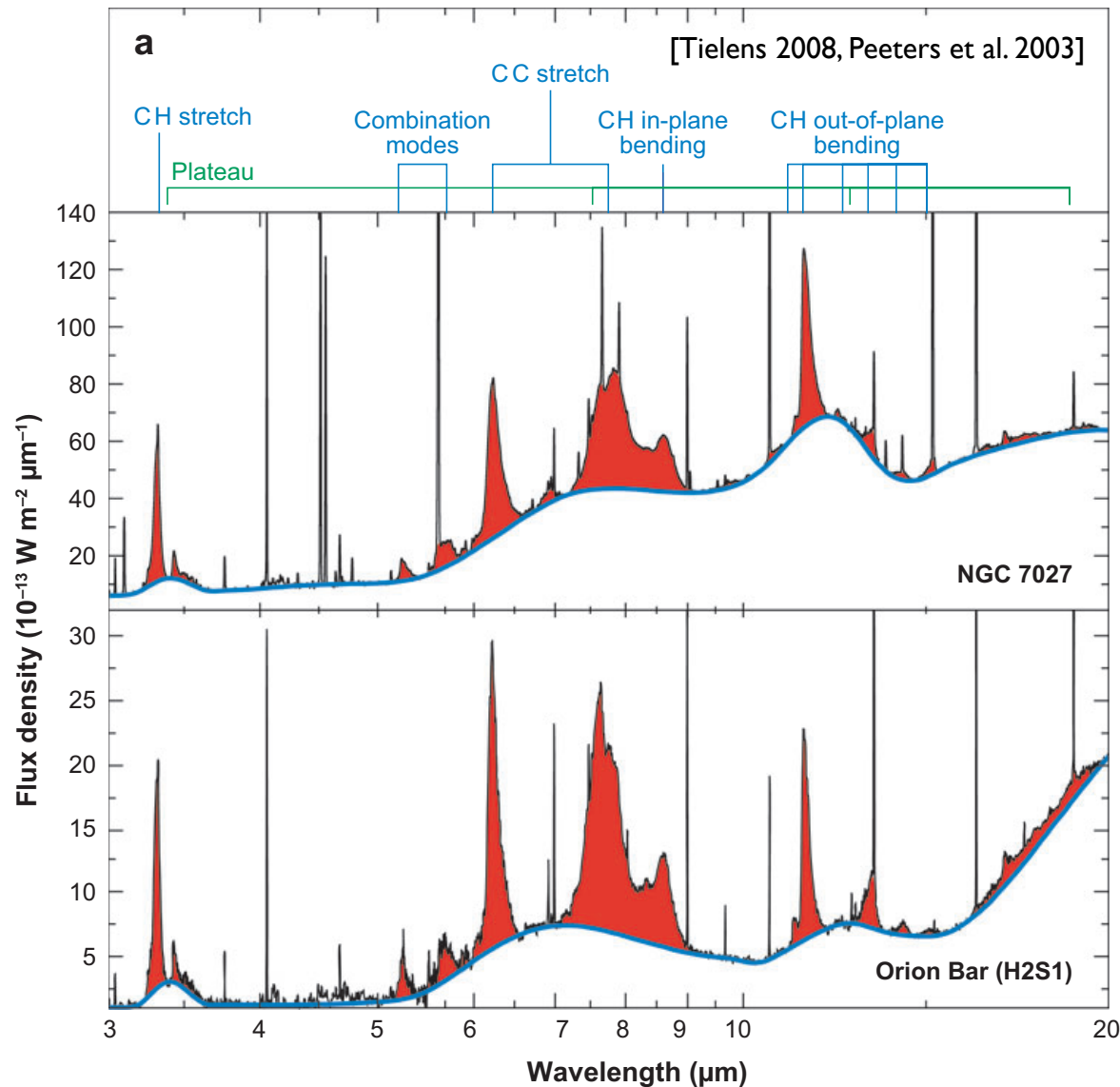
NGC 7023, 8 μm



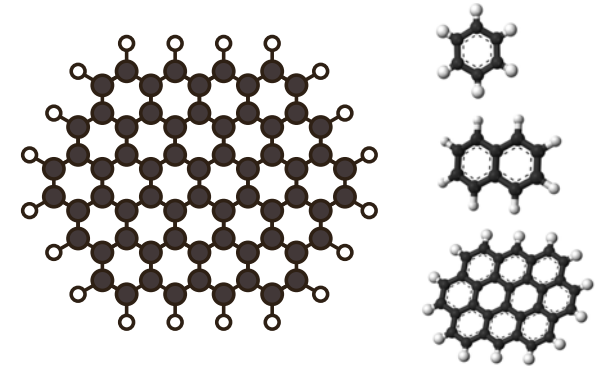
The PAH model

1984/1985 proposal that mid-IR bands are due to gas phase PAHs

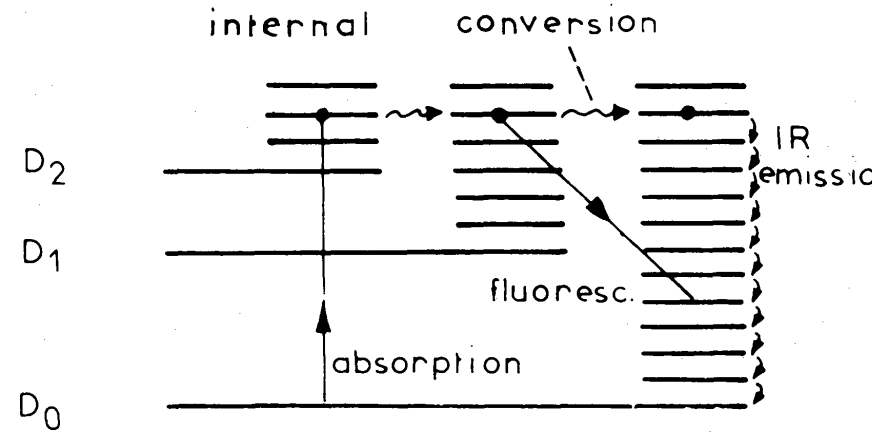
[Léger & Puget 1984] [Allamandola, Tielens, Barker 1985]



Polycyclic Aromatic Hydrocarbons (PAHs)



Emission by fluorescence (gas phase)



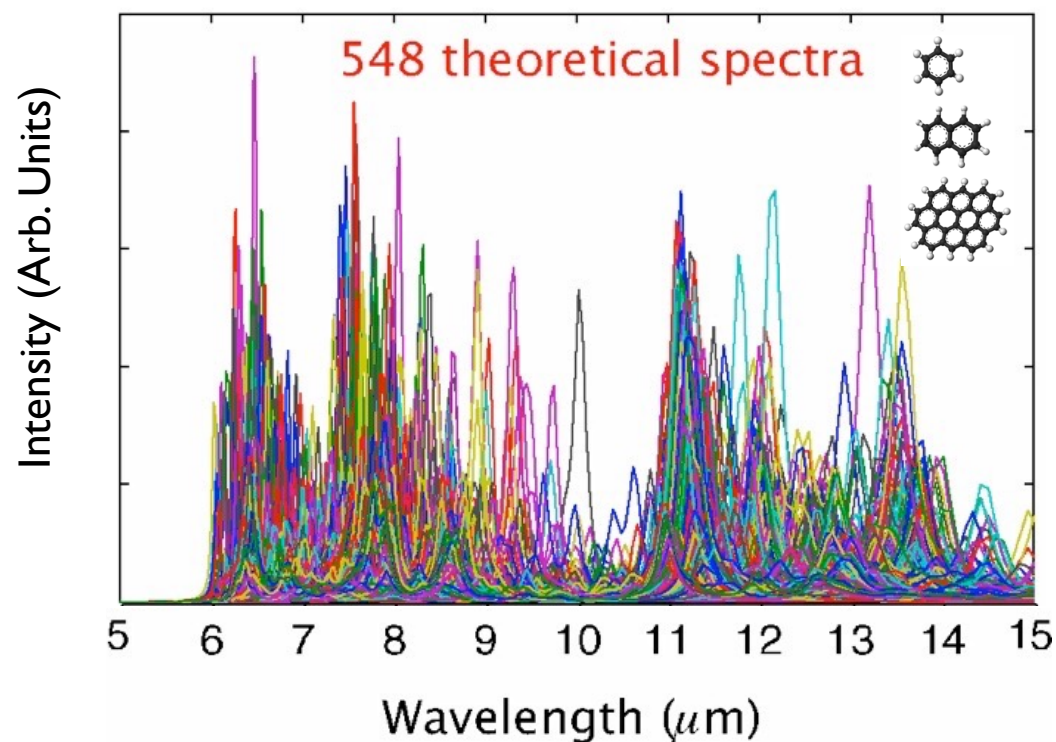
[Léger, d'Hendecourt, Défourneau, 1989]

No specific PAH molecule identified !

Random mixtures of PAHs

[Rosenberg, Berné, Boersma A&A 2014]

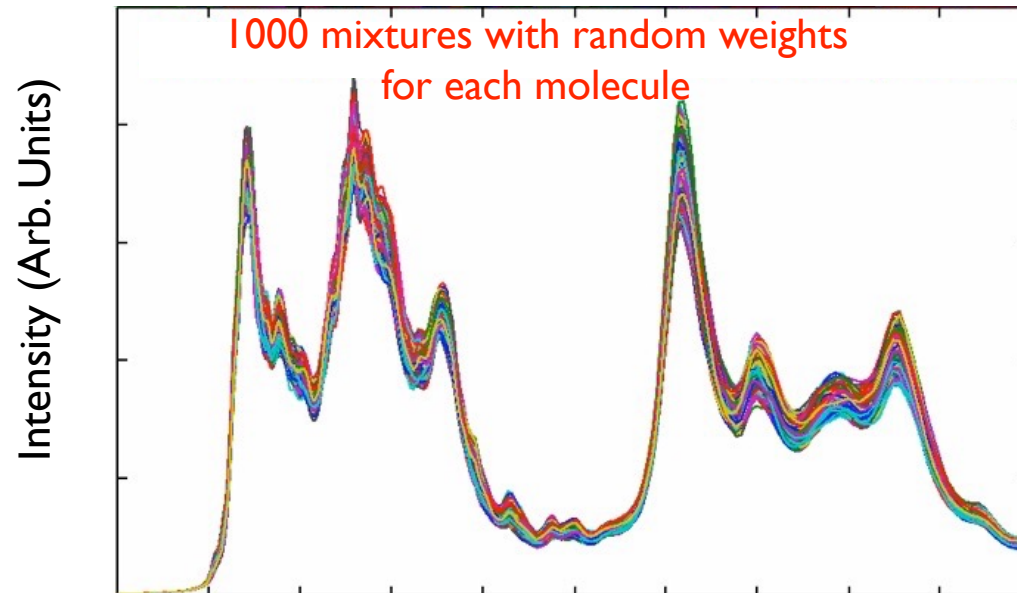
The NASA-Ames database: calculated spectra for more than 600 species, with a range of size and charge state <http://www.astrochem.org/pahdb/> [Bauschlicher, Boersma et al. 2012] [Boersma et al. 2014]



Random mixtures of PAHs

[Rosenberg, Berné, Boersma A&A 2014]

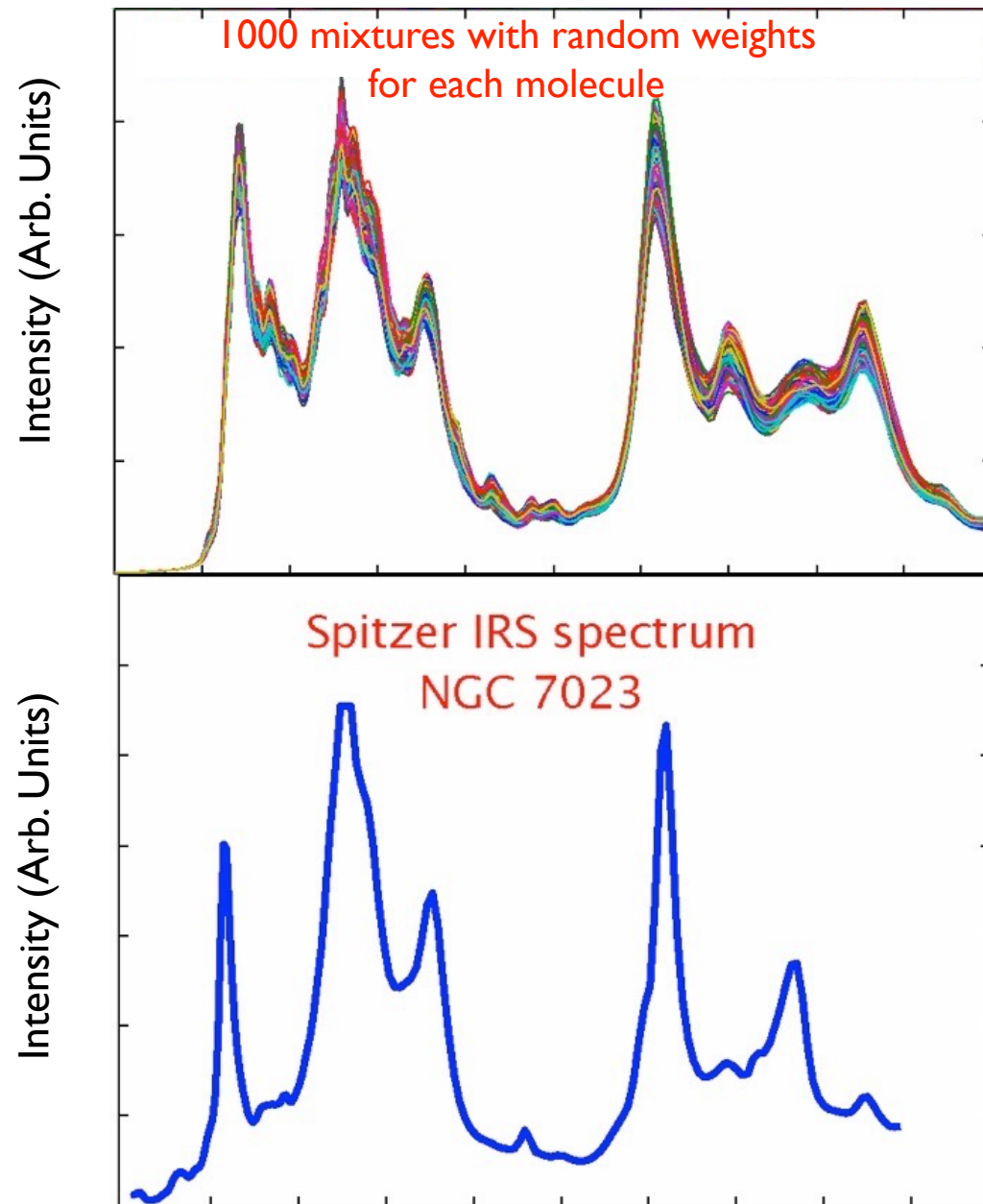
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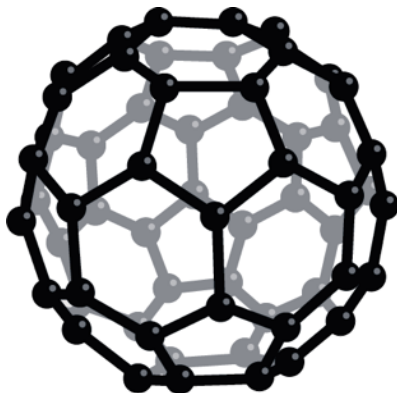
Random mixtures of PAHs

[Rosenberg, Berné, Boersma A&A 2014]

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The serendipitous discovery of C₆₀



1985 Discovery of the C₆₀ molecule in the lab

[Kroto, Heath, O'Brien, Curl, Smalley, Nature, 1985]

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LETTERS TO NATURE

NATURE VOL. 318 14 NOVEMBER 1985

C₆₀: Buckminsterfullerene

H. W. Kroto*, J. R. Heath, S. C. O'Brien, R. F. Curl
& R. E. Smalley

Rice Quantum Institute and Departments of Chemistry and Electrical Engineering, Rice University, Houston, Texas 77251, USA

During experiments aimed at understanding the mechanisms by which long-chain carbon molecules are formed in interstellar space and circumstellar shells¹, graphite has been vaporized by laser irradiation, producing a remarkably stable cluster consisting of 60 carbon atoms. Concerning the question of what kind of 60-carbon atom structure might give rise to a superstable species, we suggest a truncated icosahedron, a polygon with 60 vertices and 32 faces, 12 of which are pentagonal and 20 hexagonal. This object is commonly encountered as the football shown in Fig. 1. The C₆₀ molecule which results when a carbon atom is placed at each vertex of this structure has all valences satisfied by two single bonds and one double bond, has many resonance structures, and appears to be aromatic.

Fig. 1 A football (in the United States, a soccerball) on Texas grass. The C₆₀ molecule featured in this letter is suggested to have the truncated icosahedral structure formed by replacing each vertex on the seams of such a ball by a carbon atom.

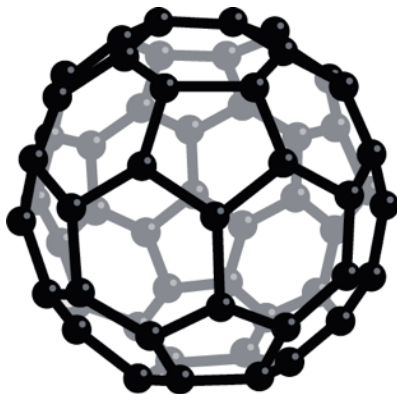


graphite fused six-membered ring structure. We believe that the distribution in Fig. 3c is fairly representative of the nascent distribution of larger ring fragments. When these hot ring clusters are left in contact with high-density helium, the clusters equilibrate by two- and three-body collisions towards the most stable species, which appears to be a unique cluster containing 60 atoms.

When one thinks in terms of the many fused-ring isomers with unsatisfied valences at the edges that would naturally arise

The serendipitous discovery of C₆₀

Richard Buckminster Fuller



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C₆₀: Buckminsterfullerene

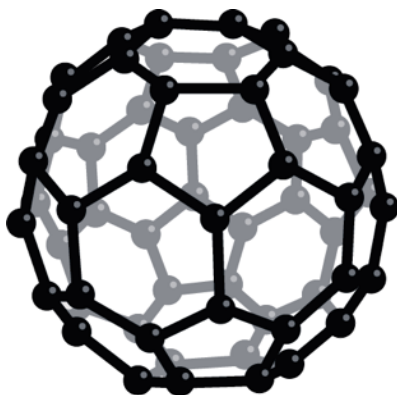
H. W. Kroto*, J. R. Heath, S. C. O'Brien, R. E. Smalley

Rice Quantum Institute and Departments of Chemistry and Engineering, Rice University, Houston, Texas 77251, U.S.A.

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The serendipitous discovery of C₆₀



162

C₆₀: Buckminsterfullerene

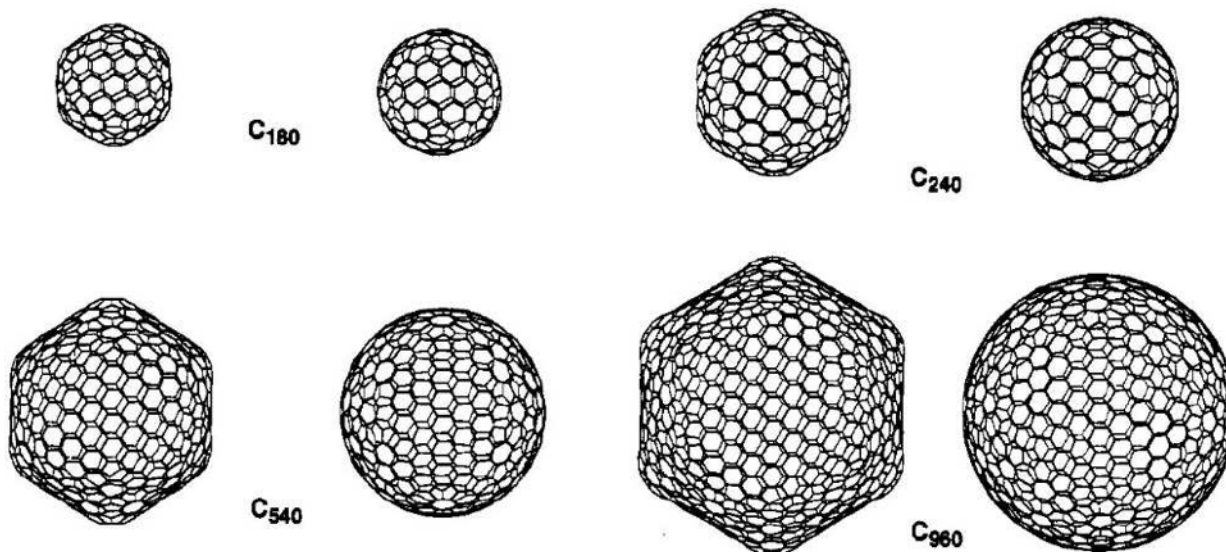
H. W. Kroto*, J. R. Heath, S. C. O'B
& R. E. Smalley

Rice Quantum Institute and Departments of C
Engineering, Rice University, Houston, Texas

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We are disturbed at the number of letters and syllables in the rather fanciful but highly appropriate name we have chosen in the title to refer to this C₆₀ species. For such a unique and centrally important molecular structure, a more concise name would be useful. A number of alternatives come to mind (for

Fullerenes



The serendipitous discovery of C₆₀



162

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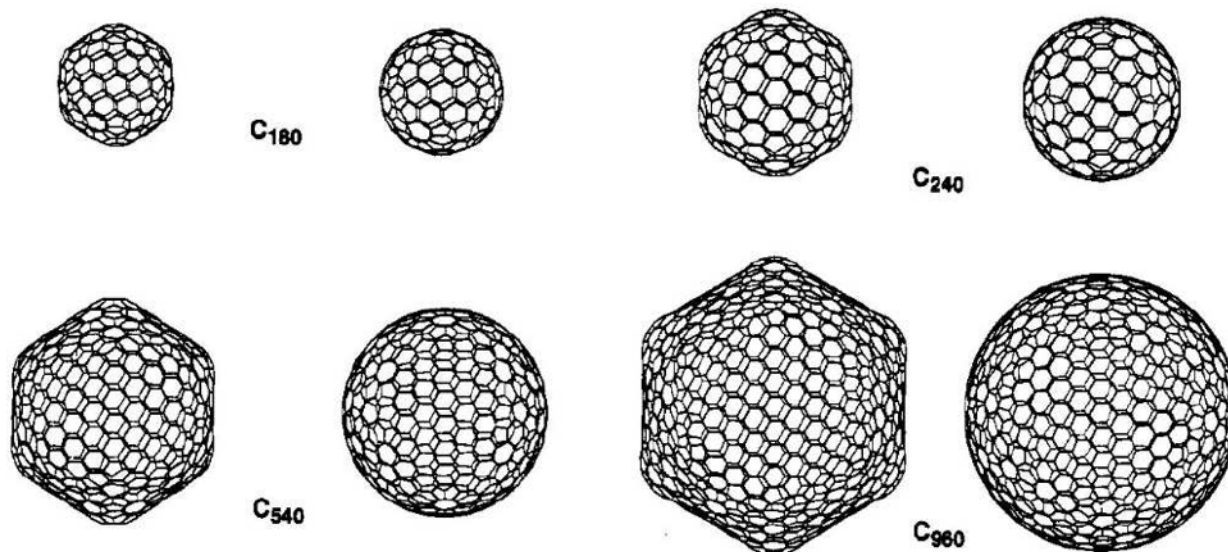
C₆₀: Buckminsterfullerene

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Fullerenes



The serendipitous discovery of C₆₀



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C₆₀: Buckminsterfullerene

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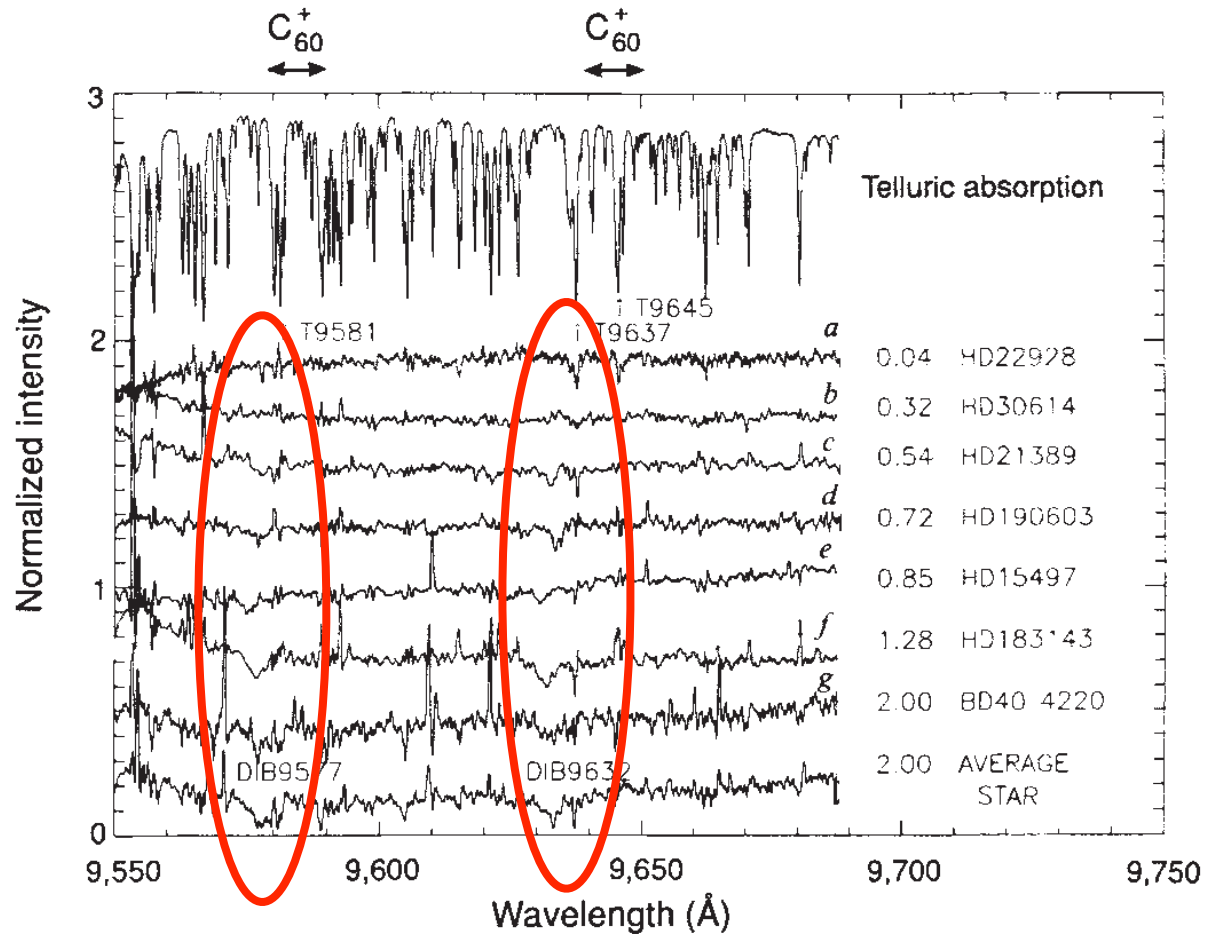
During experiments aimed at understanding the mechanisms by which long-chain carbon molecules are formed in interstellar space and circumstellar shells¹, graphite has been vaporized by laser

C₆₀ in space ?

1994

Detection of two interstellar absorption bands coincident with spectral features of C₆₀⁺

B. H. Foing* & P. Ehrenfreund†

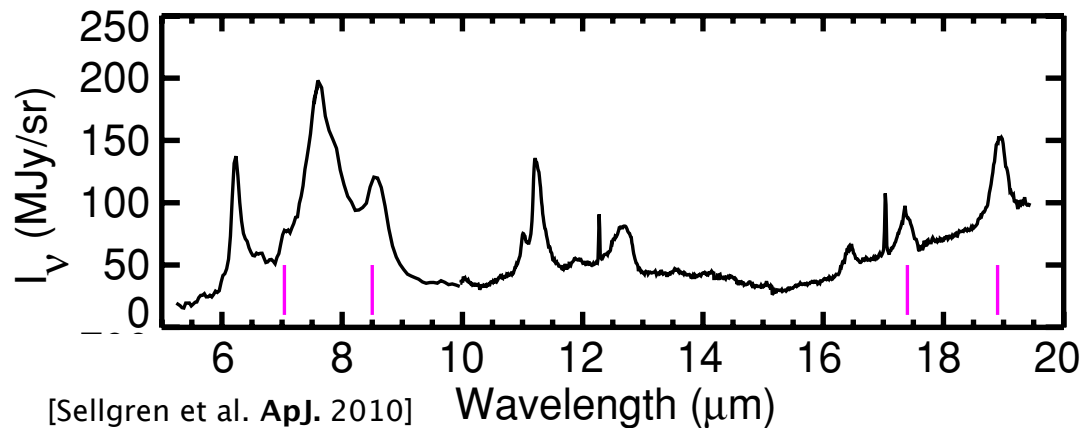


The identification of C₆₀⁺ based on coincidence of absorption bands with lab spectra is questioned by the fact that no cold, gas phase, spectroscopic data exist for this molecule

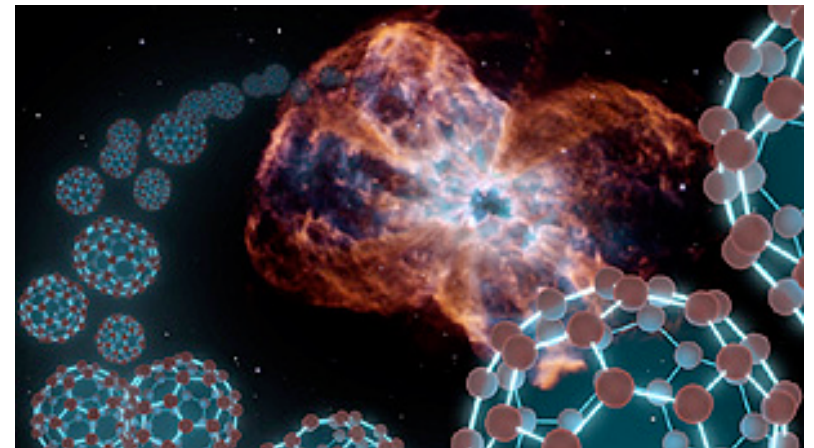
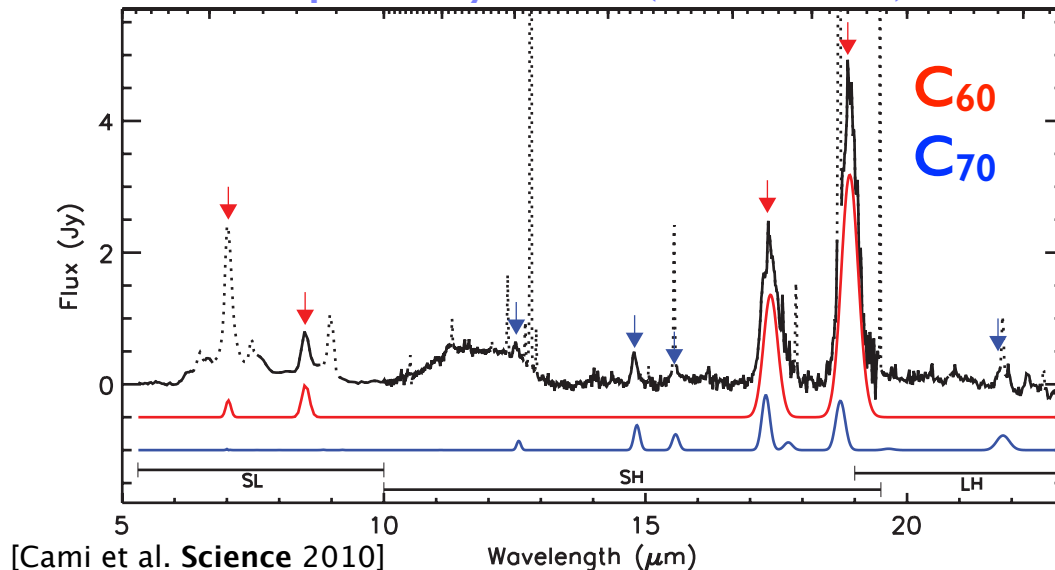
The existence of large gas phase carbon molecules

2010 Discovery of the C₆₀ molecule in emission in space

NGC 7023 reflection nebula (interstellar medium)



Tc1 planetary nebula (evolved star)



Neutral form ! + Proposed to be in the solid phase by Cami et al. while Gas phase by Sellgren et al.

2010 summary of all the problems

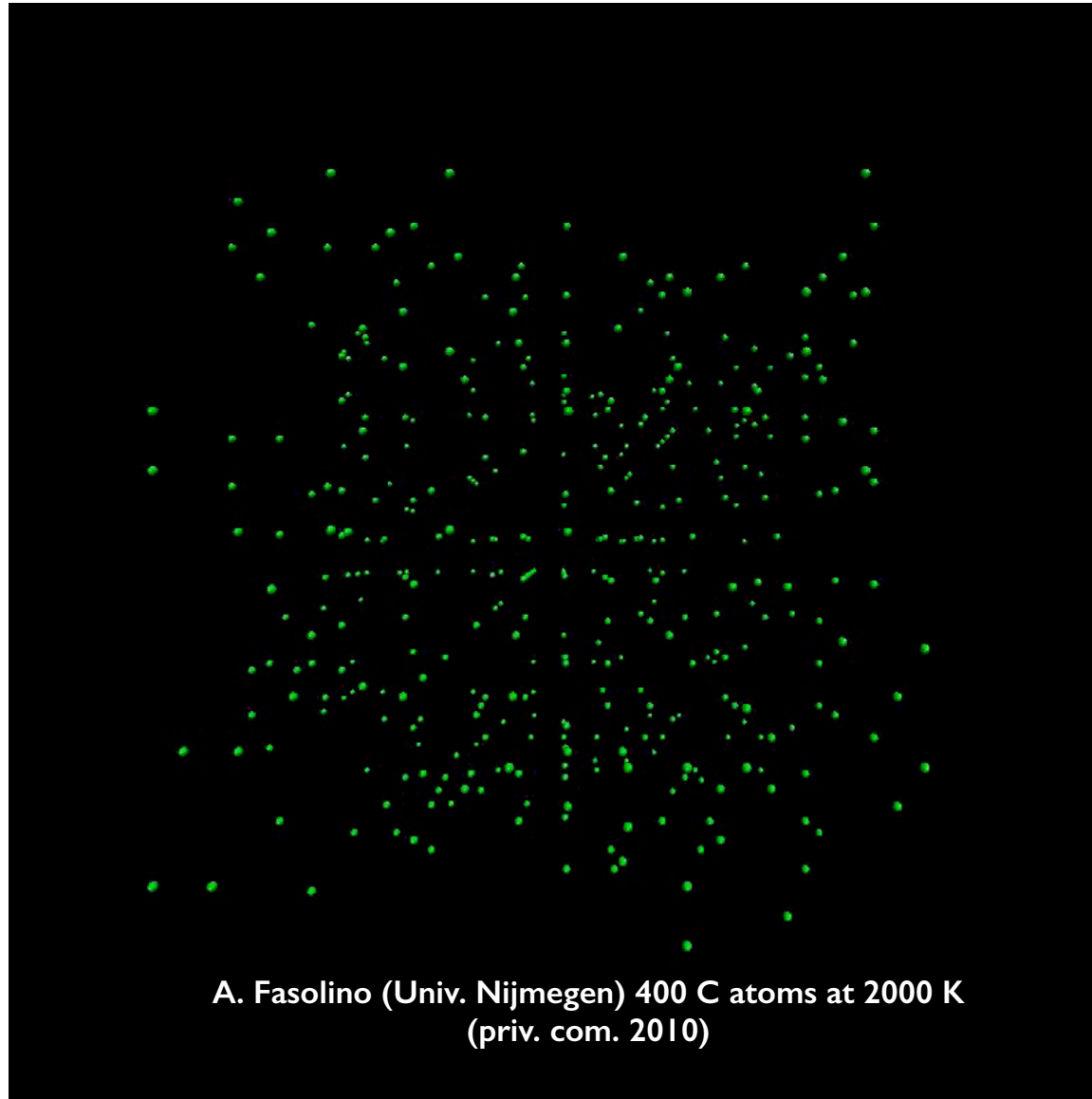
On C₆₀

- How is C₆₀ formed ?
- C₆₀ was detected, but it could not be proven that it is fluorescing
- C₆₀ was detected, but given the harsh conditions in the ISM it should be ionized, in fact there is a claim of detection of C₆₀⁺ in absorption...

On the PAH model

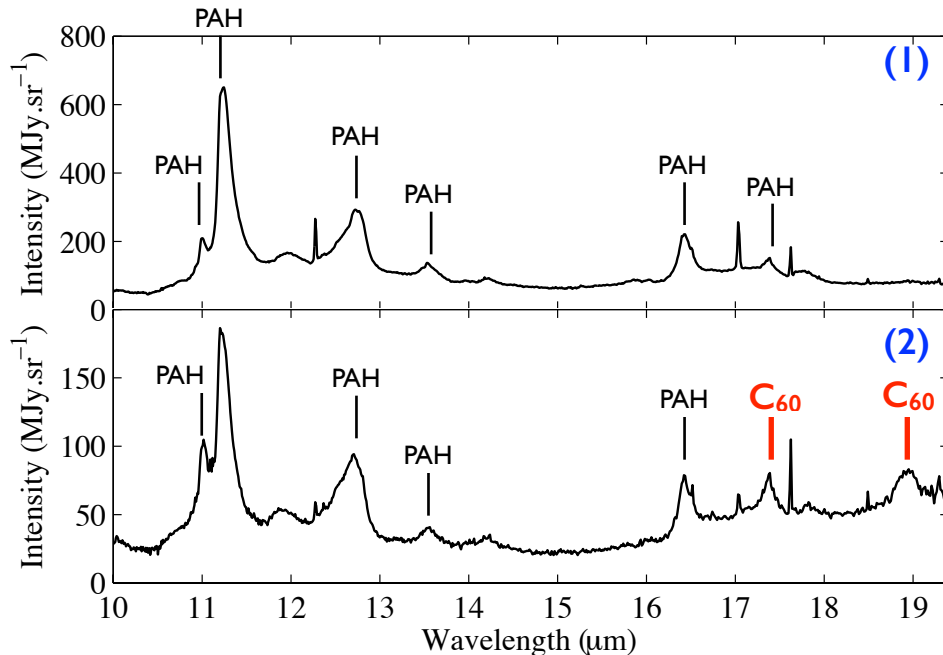
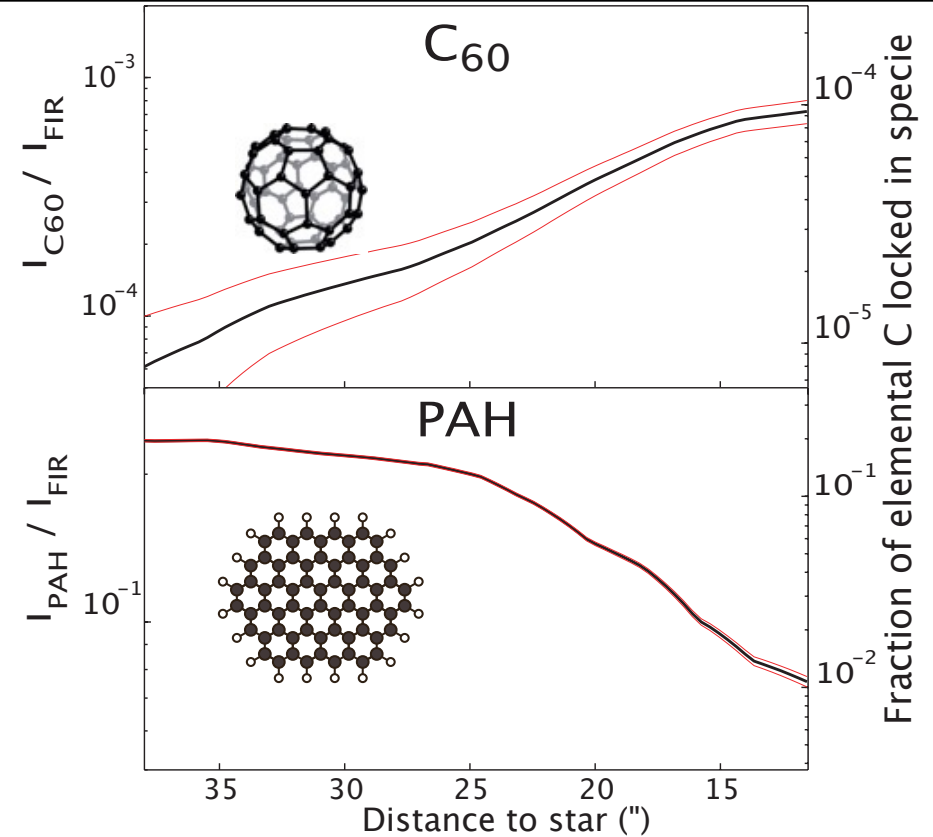
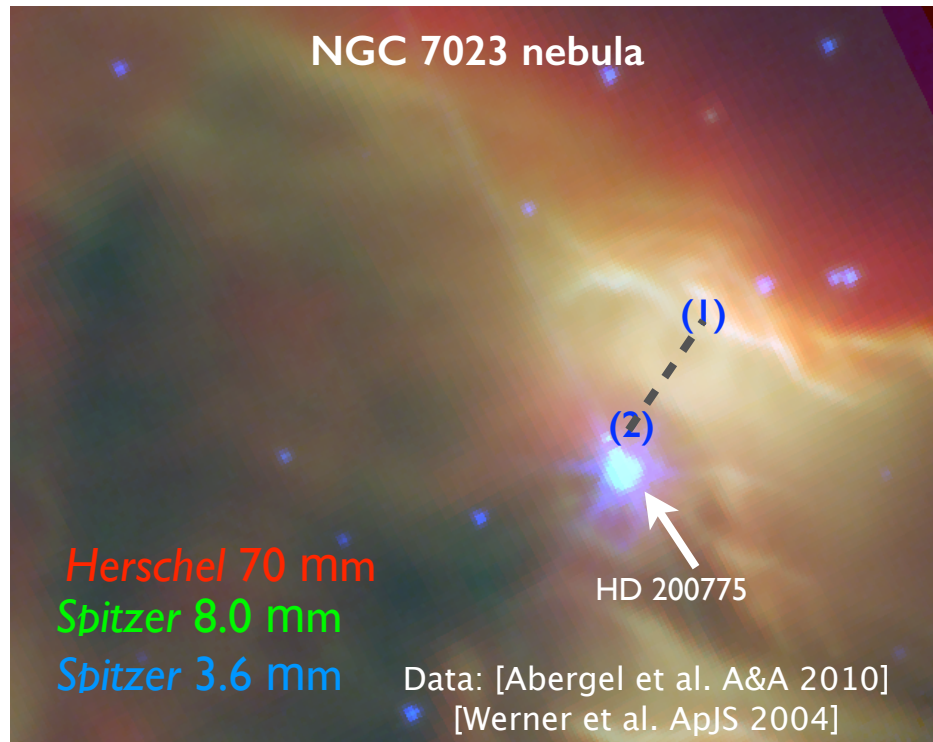
- The definitive proof of the existence of a specific gas phase carbon macro-molecule, heated to high T by individual UV photon has not been provided yet... for many astronomers this is a severe caveat of the PAH model (see dispute in the literature recently : Kwok & Zhang vs Li & Draine)

The formation of C_{60}



[Smalley 1992, Goroff 1996, Kroto & McKay 1988, Heath 1991, Hunter et al. 1992, Rubin et al. 1991, Irle et al. 2006, Huang et al. 2007, Dunk et al. 2013]

Observational evidence of C₆₀ formation in the ISM



C₆₀ is formed in the interstellar medium, at low density ($n_H = 100 \text{ cm}^{-3}$) !

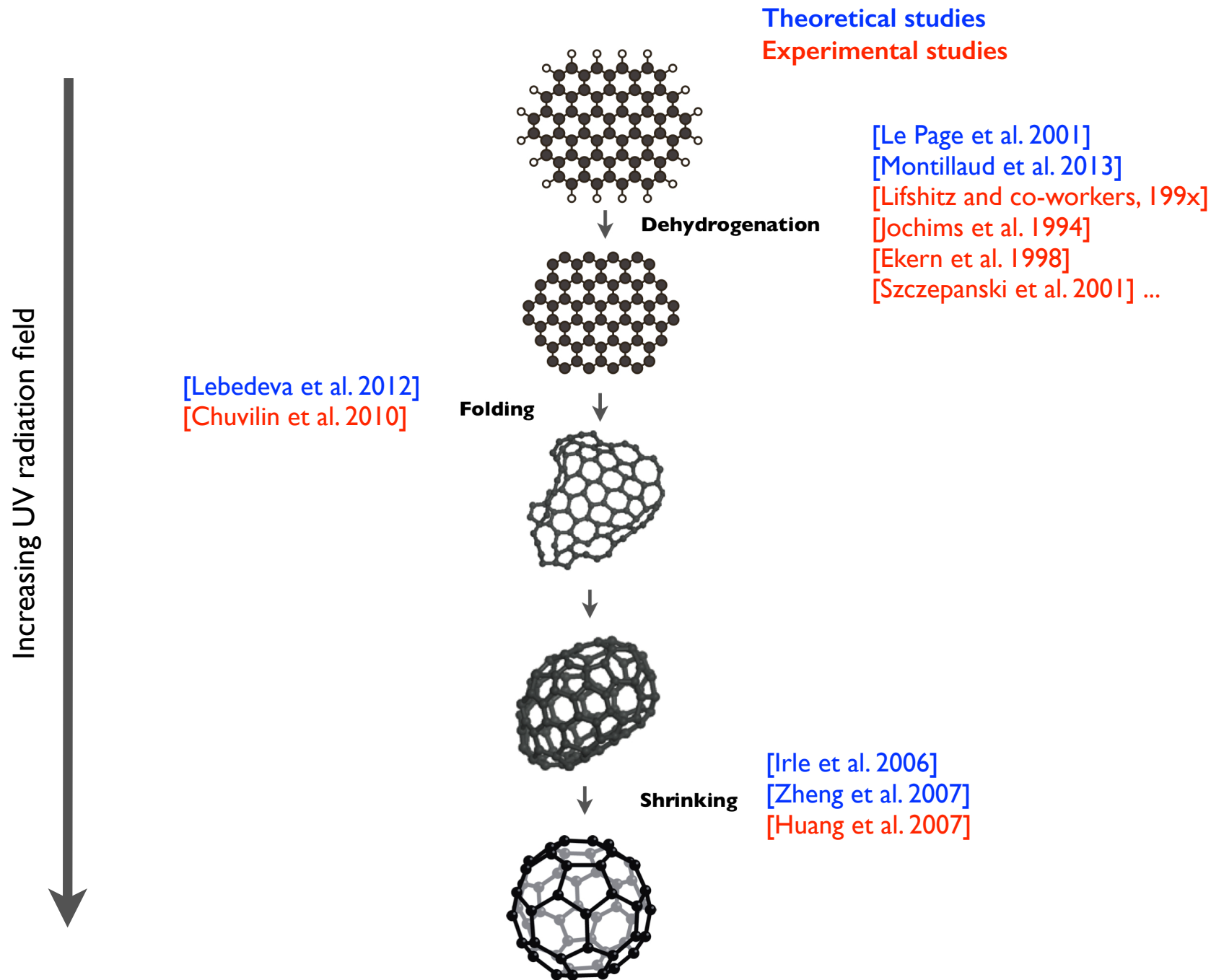
Aggregation (bottom up) process not possible !

C₆₀ formation from photochemical PAH processing ?

Conversion efficiency from PAHs to C₆₀ ~0.1%
at 15'' from the star in 10⁵ years (age of the nebula)

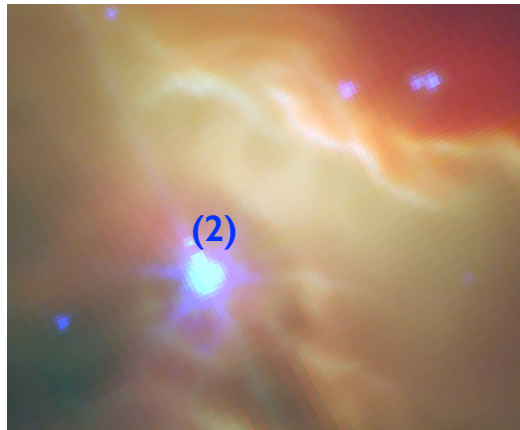
[Berné & Tielens PNAS 2012]

Proposed scenario

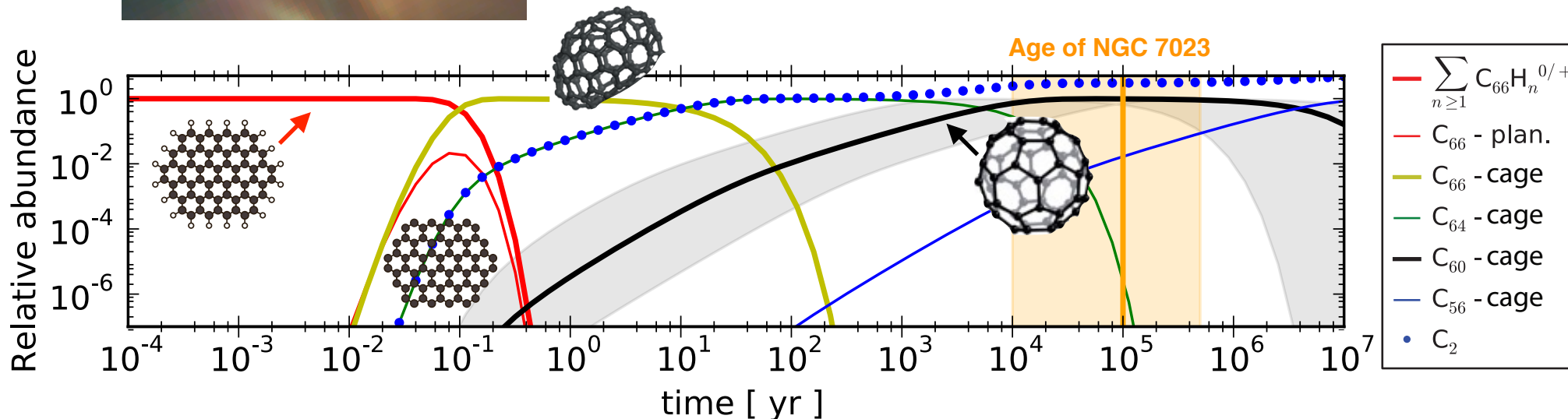


Photochemical model for $C_{66}H_{20}$ to C_{60} in NGC 7023

[Berné, Montillaud, Joblin, A&A 2015]

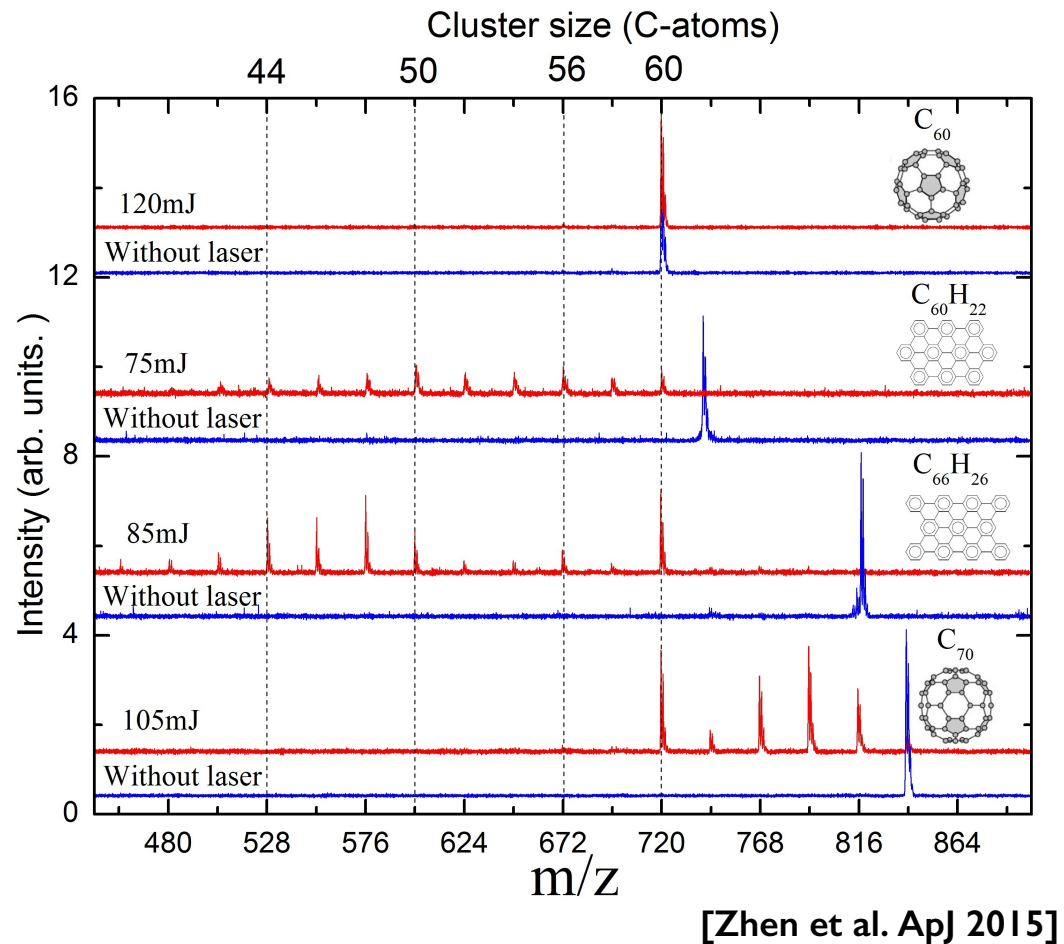


At 10'' from the star
 $G_0 \sim 10^5$
 $n_H \sim 500$



- Circumovalene quickly dehydrogenated
- Graphene flakes (dehydrogenated PAHs) are unstable,
- Shrinking step takes typically 10^4 years,
- Yet we can reach C_{60} in about $\sim 10^5$ years a timescale comparable to the age of NGC 7023,
- C_2 molecules are formed from the destruction of cages but will likely be destroyed,
- Similarly, PAHs smaller than 60 C atoms will be destroyed (unless converted in something more stable),
- In the end, **only C_{60}** survives to this top down chemistry...

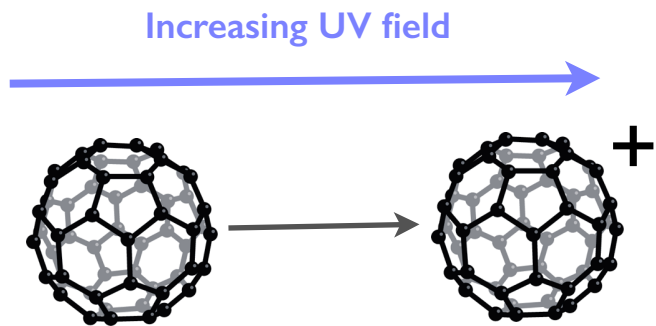
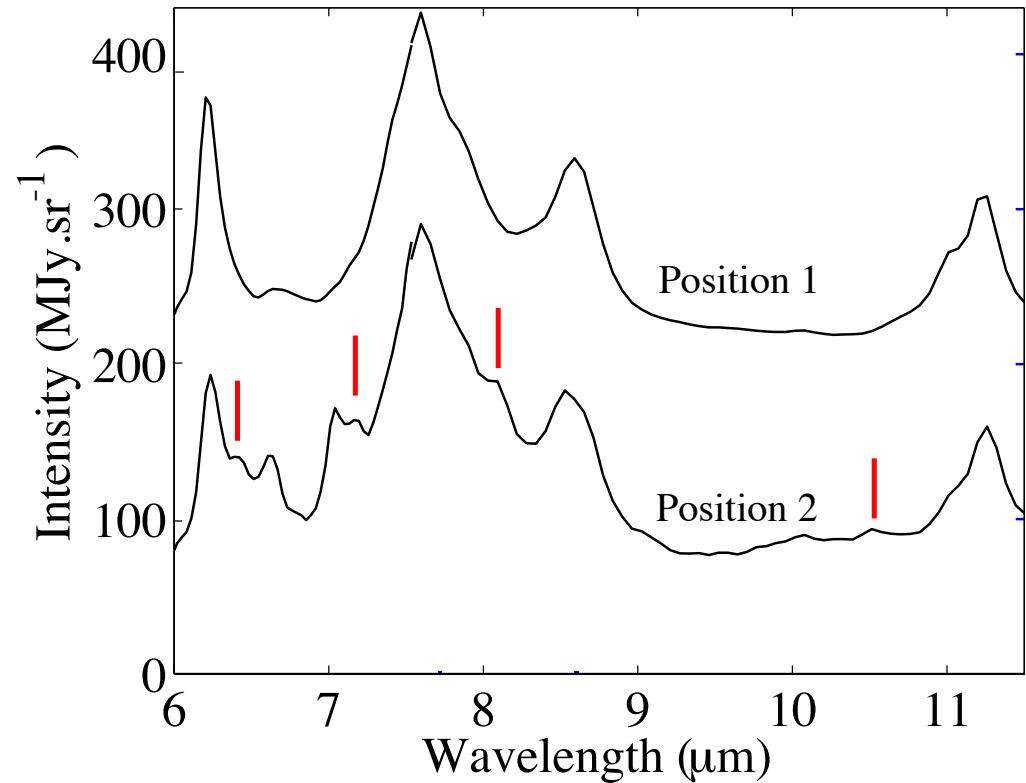
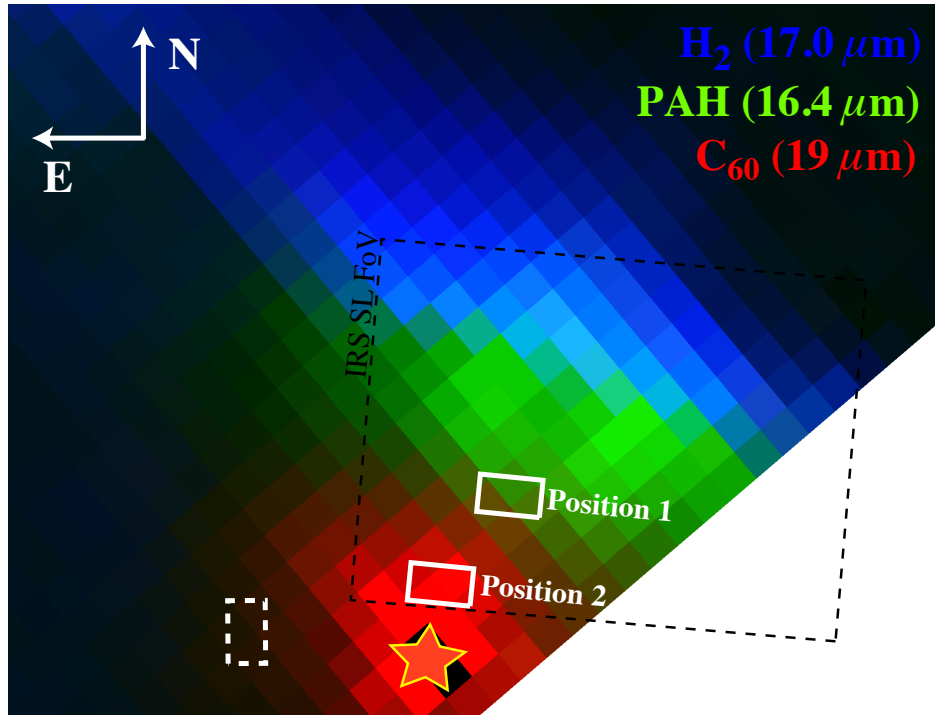
Laboratory confirmation



- Top down formation of C_{60} from PAHs confirmed in the laboratory
- However, pathway leading to C_{60} not clear, in particular the sequence folding vs C_2 loss

New bands in NGC 7023

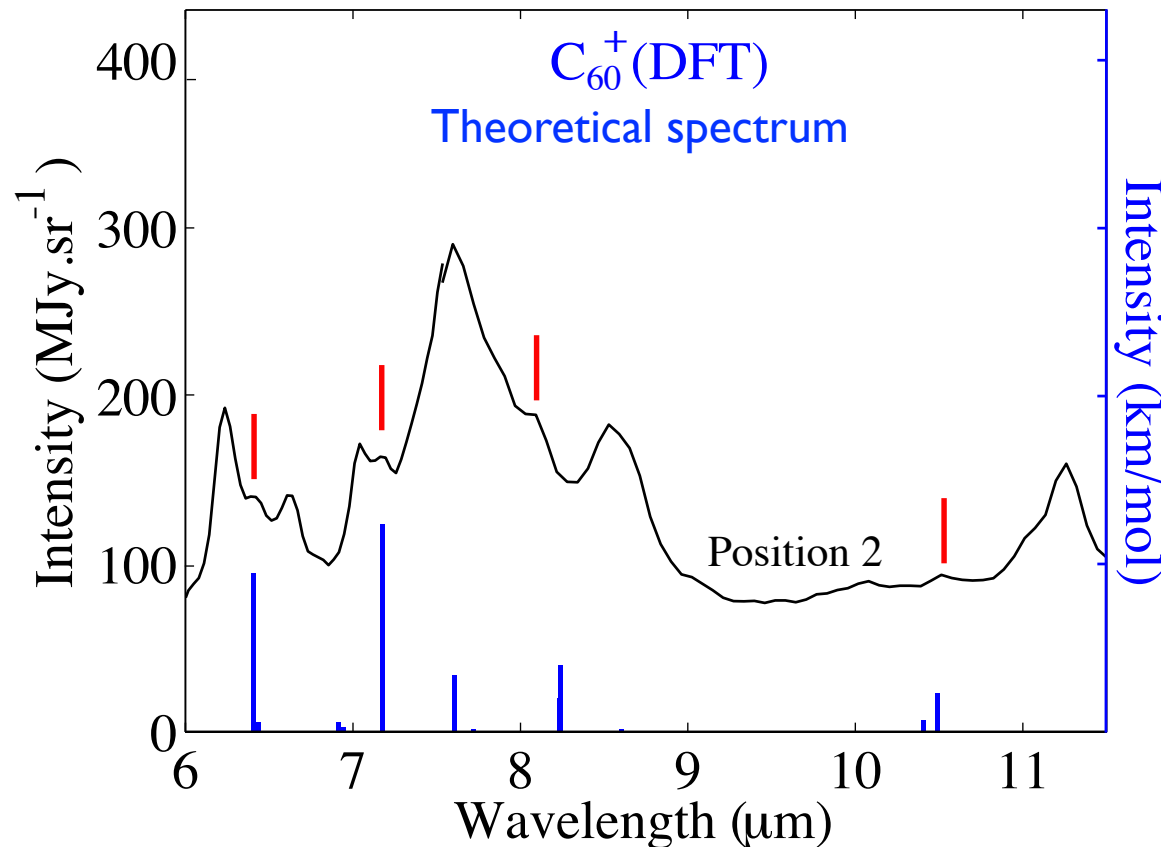
Spitzer IRS long wavelength



- Bands at 6.4, 7.1, 8.3 and 10.5 μm
- These bands are only present near the star
- Neutral C_{60} emission is more extended

Spectroscopy of C_{60}^+ : theory vs observation

[Berné, Mulas, Joblin 2013]



Even without any emission model (which even for C_{60} is not available):

- Harmonic spectrum matches observed positions with 2% accuracy
- Bands with strongest intensities are strongest in observation

Ionization fraction and abundance:

- At observed position 38% of C_{60} is ionized
- Abundance : max 10^{-4} of the elemental C would be in C_{60}^+ (much smaller than F&E)

2015 remaining problems

On C₆₀ fullerene

- C₆₀ was detected, but it could not be proven that it is fluorescing
- The detection of C₆₀⁺ suggests it is fluorescence
- C₆₀ was detected, but given the harsh conditions in the ISM it should be ionized, in fact there is a claim of detection of C₆₀⁺ in absorption...
- We detected the emission of C₆₀⁺
- How is C₆₀ formed ?
- We suggest it is formed from the photoprocessing of PAHs and it is confirmed in the lab that this is possible

On the PAH model

- The definitive proof of the existence of a specific gas phase carbon macro-molecule, heated to high T by individual UV photon has not been provided yet... for many astronomers this is a severe caveat of the PAH model
- C₆₀⁺ is a counter example

Rugbene

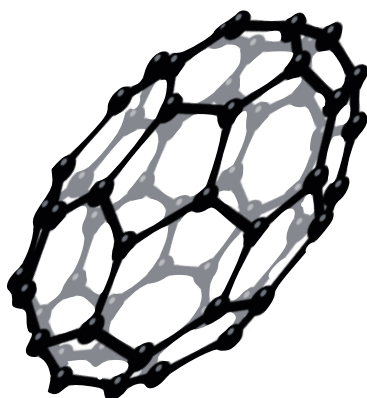
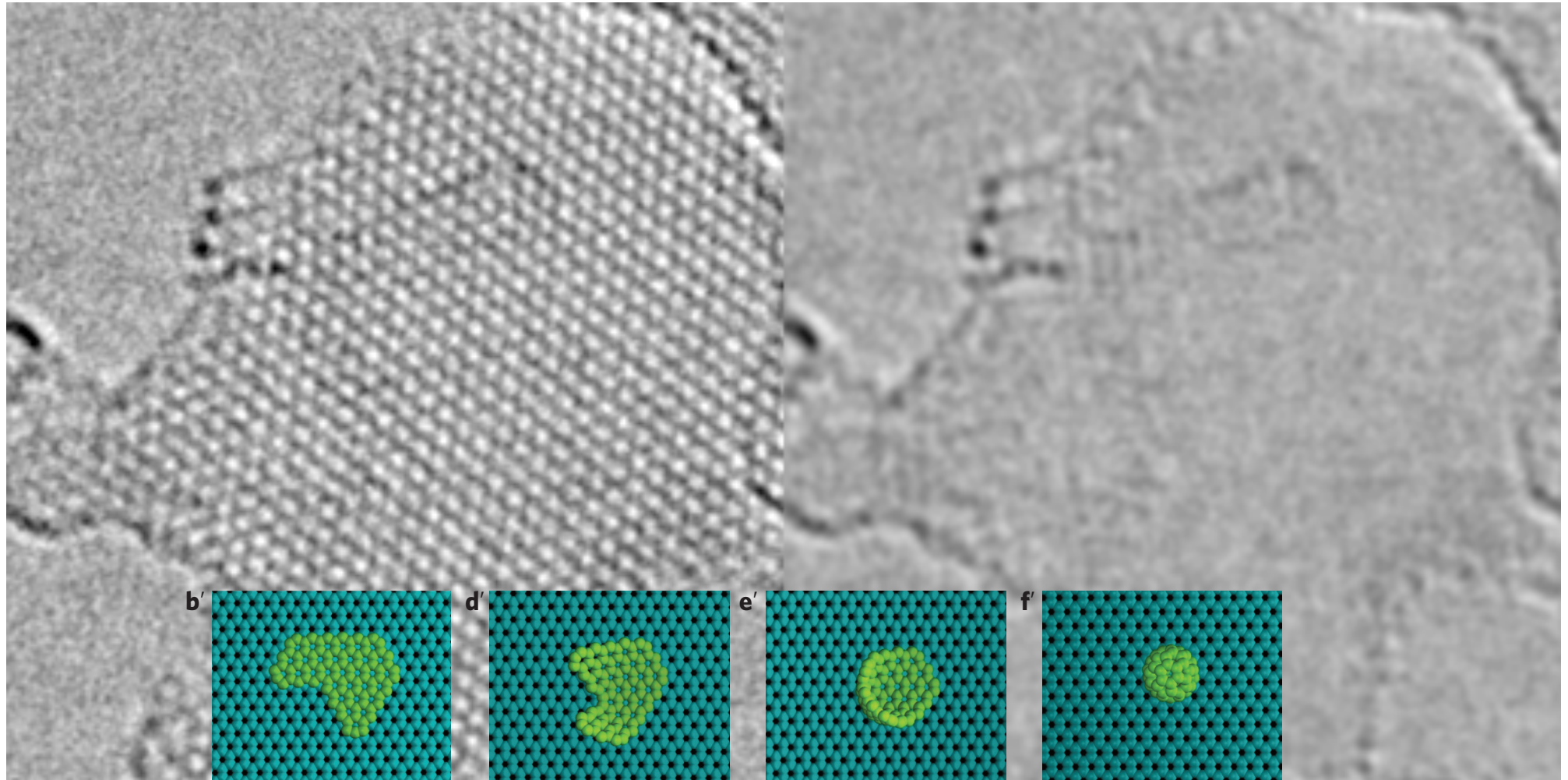


Fig. 1. a rugby ball on Toulouse grass.



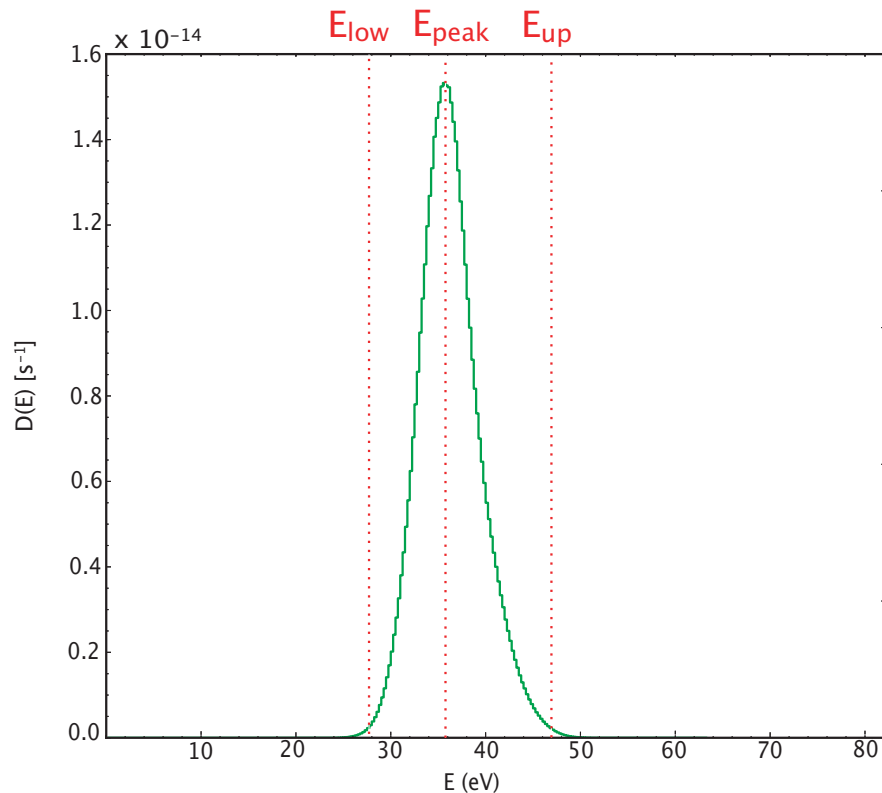
The roll up route

[Chuvilin et al. Nature Chem. 2010]



The remarkable stability of C₆₀

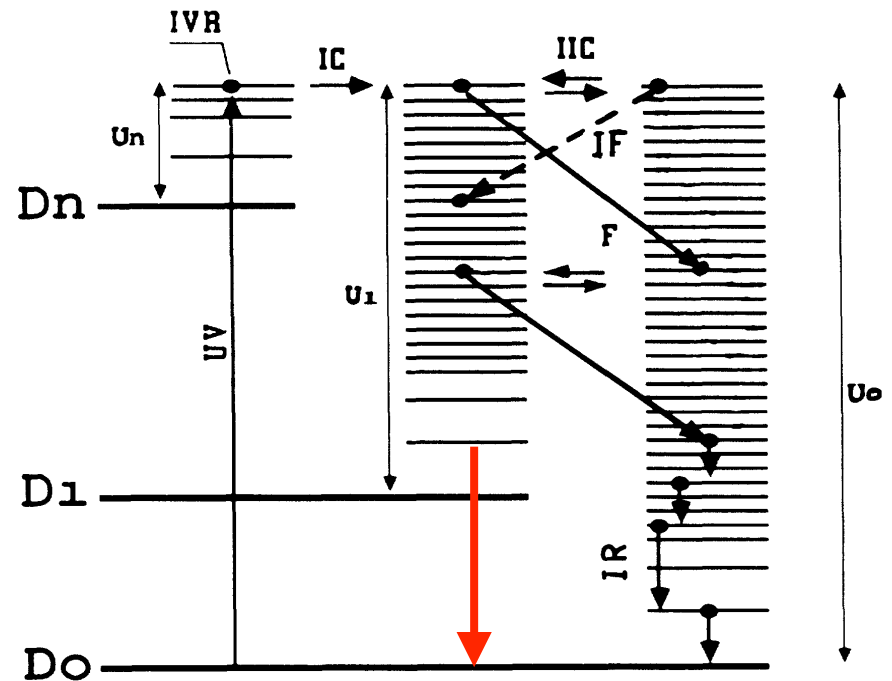
Probability of dissociation of C₆₀ at 5'' from the star, $G_0=2 \times 10^5$



Energies of dissociation for the cages

a.			
Species*	E_{low}	E_{peak}	E_{up}
C ₆₆	22	32	39
C ₆₄	24	35	43
C ₆₂	15	20	28
C ₆₀	27	36	47
C ₅₈	22	30	37

The energy required to break the cages is huge, hence these systems can survive at very high internal energies. In this case, emission from thermally excited electronic states a.k.a. **Poincaré Fluorescence** becomes important.



[Léger PRL 1988]